

MT 2 [2023-24] ANSWER KEY

Max. Marks- 20

Time:50 minutes

Q.No.	Answers	Mark
	Section A	
1	С	1
2	В	1
3	В	1
4	В	1
5	С	
	Section B	
6	Chemical formulas are used to describe the different types of atoms and their numbers in a compound or element. Each element's atoms are symbolised by one or two letters. A collection of chemical symbols that depicts the elements that make up a compound and their quantities. (a) Magnesium chloride – MgCl ₂ (b) Calcium oxide – CaO (c) Copper nitrate – Cu(NO ₃) ₂ (d) Aluminium chloride – AlCl ₃ (e) Calcium carbonate – CaCO ₃	2
7	Force applied	
0	Object must be displaced	2
0	P=W/C P=1000/10=100i	
9	$\begin{array}{l} \text{KE}=1/2\text{mv}^2 \\ \text{25}=1/2 \text{ m} 5^2 \\ \text{M}=2 \text{ kg} \\ \text{When } \text{v}=10 \text{ m/s} \\ \text{KE}=100\text{j} \\ \text{When } \text{v}=15 \text{ m/s} \\ \text{KE}=225 \text{ j} \end{array}$	-
10	The molecular mass of a substance is the sum of the atomic masses of all the atoms in a molecule of the substance. It is expressed in the atomic mass unit(u). The molecular mass of $H_2 - 2 x$ atoms atomic mass of $H = 2 x 1u = 2u$ The molecular mass of $O_2 - 2 x$ atoms atomic mass of $O = 2 x 16u = 32u$	

	The molecular mass of $Cl_2 - 2 \times atoms atomic mass of Cl = 2 \times 35.5 u$	
	= 71u	
	The molecular mass of CO_2 – atomic mass of C + 2 x atomic mass of	2
	$O = 12 + (2 \times 16)u = 44u$	
	The molecular mass of CH_4 – atomic mass of C + 4 x atomic mass of	
	$H = 12 + (4 \times 1)u = 16u$	
	The molecular mass of $C_2H_2 = 2x$ atomic mass of $C_2 + 6x$ atomic mass	З
	of H = $(2 \times 12) \pm$	5
	$(6 \times 1)u - 24 + 6 - 20u$	
	$(0 \times 1)u = 24 + 0 = 30u$	
	The molecular mass of C_2H_4 - 2 x atomic mass of C + 4 x atomic mass	
	of $H = (2x \ 12) +$	
	$(4 \times 1)u = 24 + 4 = 28u$	
	The molecular mass of NH_3 – atomic mass of $N + 3 \times A$ atomic mass of	
	$H = (14 + 3 \times 1)u = 17u$	
	The molecular mass of CH ₃ OH – atomic mass of C + 3x atomic mass	
	of H + atomic mass of O + atomic mass of H = (12 +	
	3x1+16+1)u=(12+3+17)u=32u	
	$\int \frac{1}{2} \frac{1}{2} \frac{1}{3} \frac{1}{1} \frac{1}{1} \frac{1}{3} \frac{1}{1} $	
11	$y^2 - y^2 - 228$	-
**	v - u - 2as	
	$S = v_2 - u_2 2a$	
	We know that .	
	Work done - Force(E) x displacement (s)	
	work done = Porce(P) ~ displacement (s)	
	$\mathbf{F} = \mathbf{ma}$	
	$W = ma \times y_2 - u_2 2a$	
	FIL II 	
	["a" gets cancelled	
	$W = m \times v_2 - u_2 2$	
	$= 1/2 m[v^2 - u^2]$	
	when , initial velocity , $u = 0$	
	$W = 1/2mv^2$	
	K.E = Work done	
	$-1/2 \text{ my}^2$	
L	_ 1/4 my	L